

# **Electron Autodetachment of Oligonucleotide Anions in the Gas Phase**

## **Allison S. Danell and Joel H. Parks**

*ROWLAND INSTITUTE AT HARVARD, 100 EDWIN LAND BLVD., CAMBRIDGE, MA 02142*

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**This poster was originally presented in large format, but it was modified for easier reading/downloading from the web.**

# INTRODUCTION

**Trapped oligonucleotide anions have been observed to undergo sequential loss of charge as a function of temperature.**

- No changes in the deconvoluted mass spectra occur with the formation of lower charge state ions, indicating that the charge loss is the result of electron autodetachment.
- Electron autodetachment rates vary as a function of temperature, charge state, and sequence.
- The variation of the rates suggest that conformational fluctuations of single-stranded oligonucleotides play a role in the decay process.

# EXPERIMENTAL

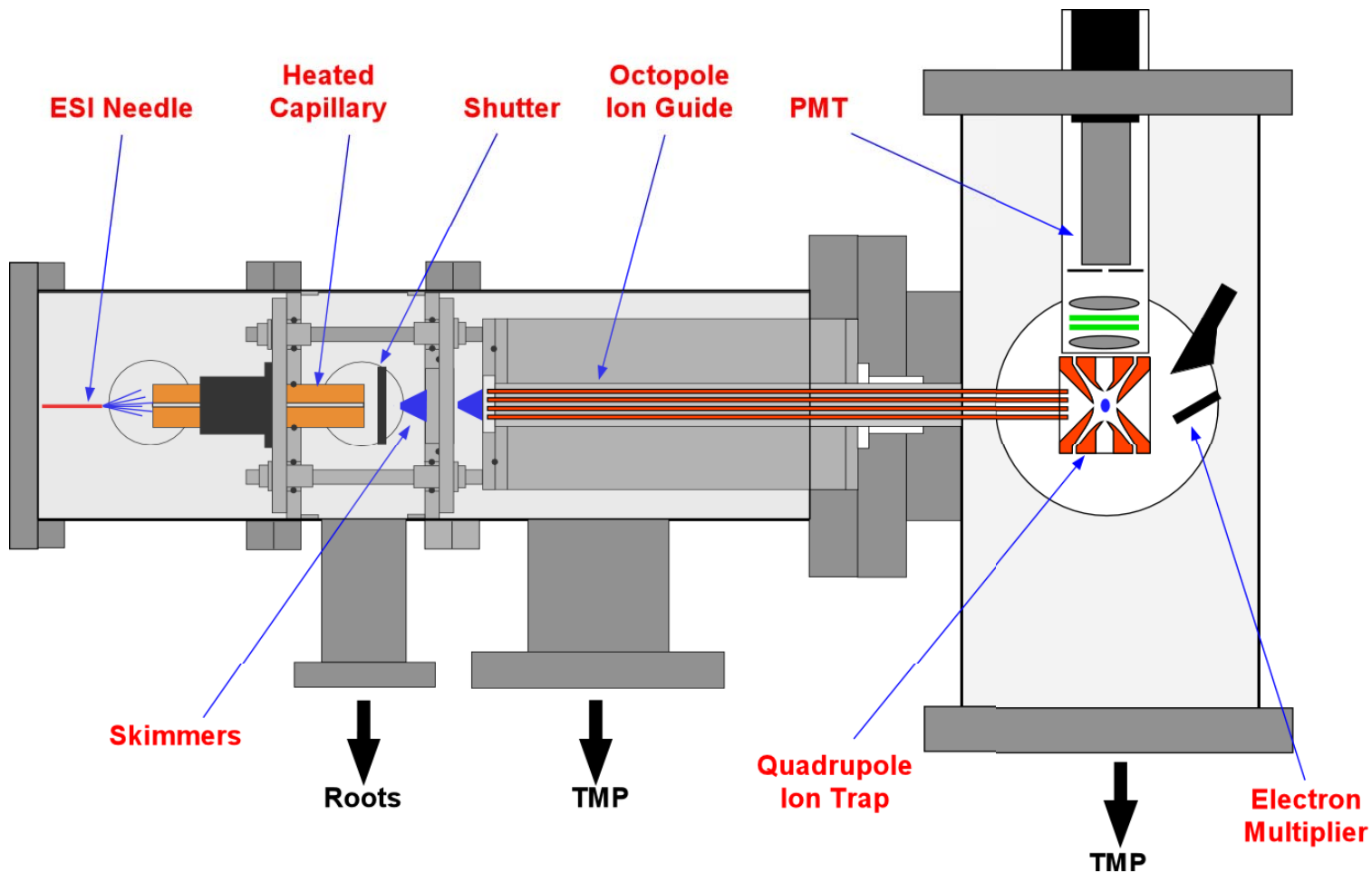
## Analytes

- Oligonucleotides purchased from Synthegen (Houston, TX)
- 5 to 10  $\mu\text{M}$  in 70/20/10 MeOH/H<sub>2</sub>O/trifluoroethanol

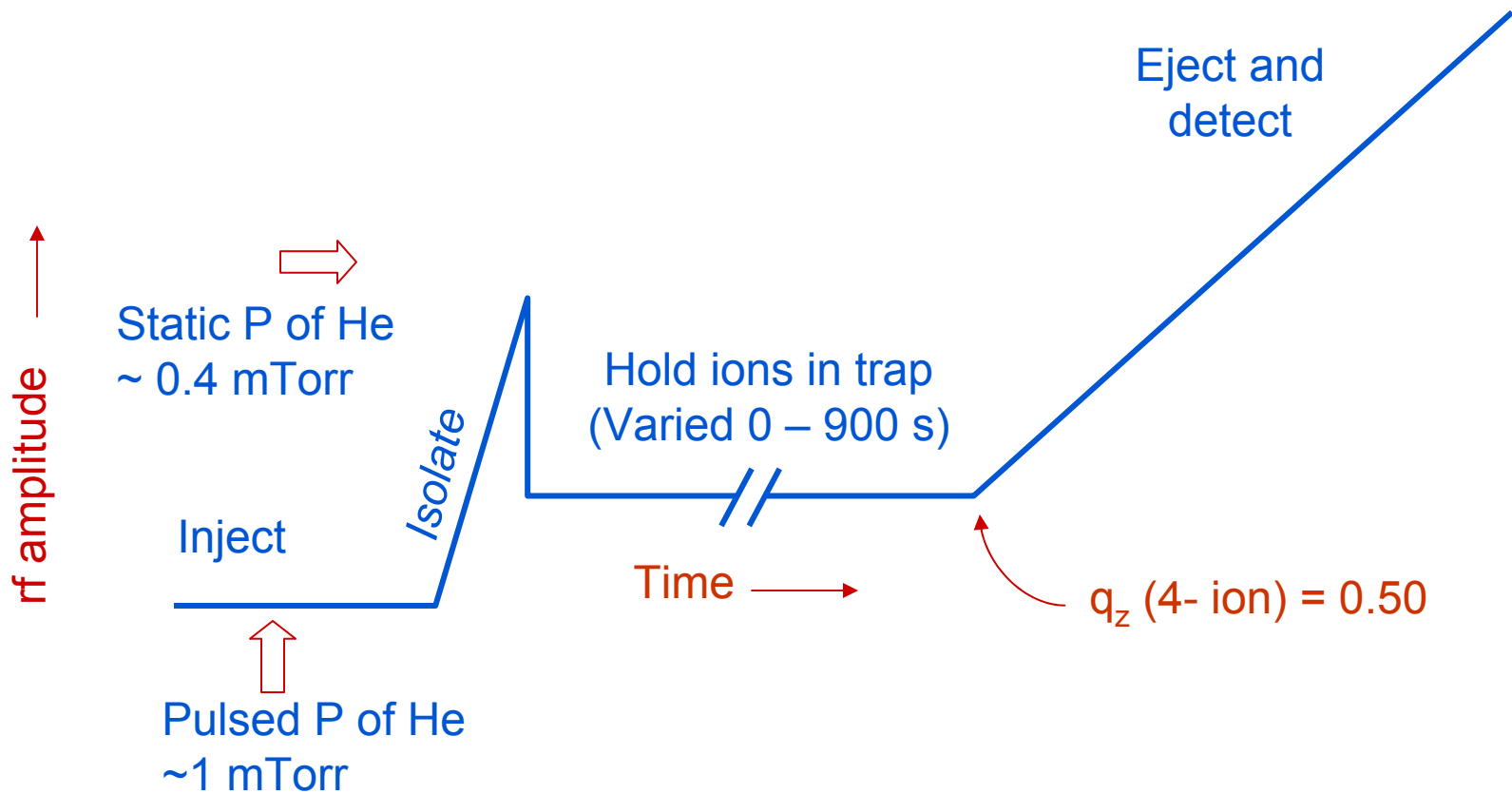
## Instrumentation

- Custom-built quadrupole ion trap
  - $\Omega = 600 \text{ kHz}$ ,  $z_0 = 3 \text{ mm}$
  - nanoESI source
- Temperature-controlled trap assembly and He background gas
  - Electrodes and He inlet seated in copper housing
    - Copper housing resistively heated with Watlow model 965 temperature controller
  - Aluminum oxide spacers
  - Heating to  $\sim 170^\circ\text{C}$
- Leybold Inficon Quadrex 200 Residual Gas Analyzer (RGA) mounted on vacuum system

# SCHEMATIC OF THE QUADRUPOLE ION TRAP ASSEMBLY



# METHODS

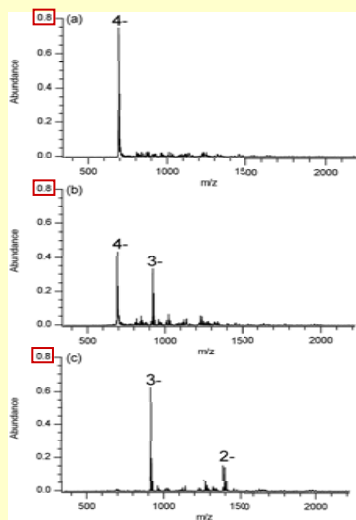


Typical scan function

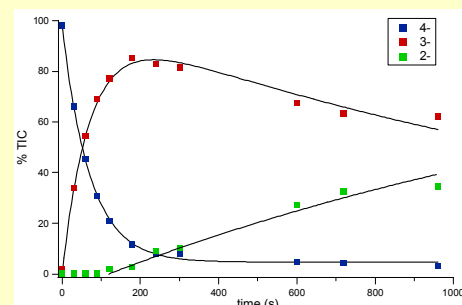
# RESULTS

## Initial observation of charge state conversion

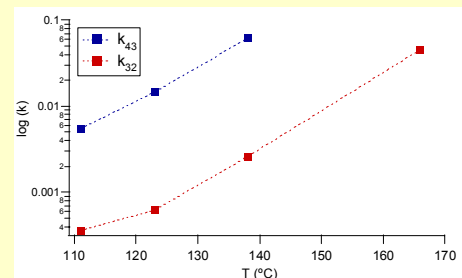
- Initially observed conversion of  $M^{n-} \rightarrow M^{(n-1)-}$  under thermal and temporal conditions used for FRET experiments [1]
  - Ion population conserved (see **absolute intensity scales** for mass spectra below)
  - Laser excitation not required
- Rate equations solved to describe first order exponential decay/growth curves observed
- Rates increased with charge state and temperature



Mass spectra of  $(T_7\text{-dye})^{4-}$  at varied heating times at 123°C. (a) 0 s (isolation of 4- charge state); (b) 30 s; (c) 960 s.



$4- \rightarrow 3- \rightarrow 2-$  for  $T_7$ -dye as  $f(t)$  at 123°C

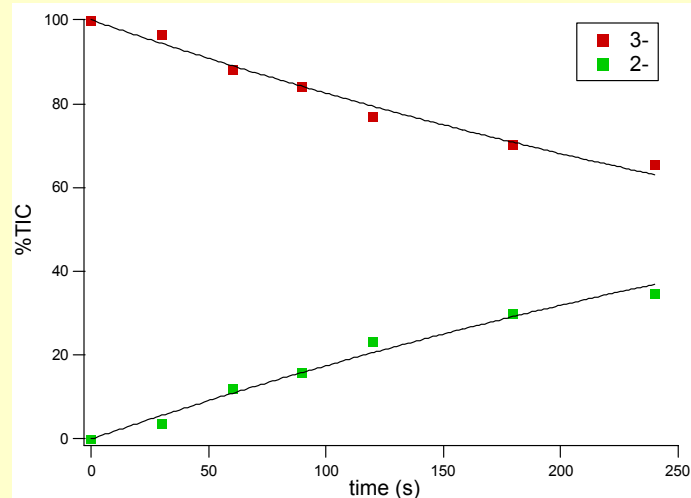
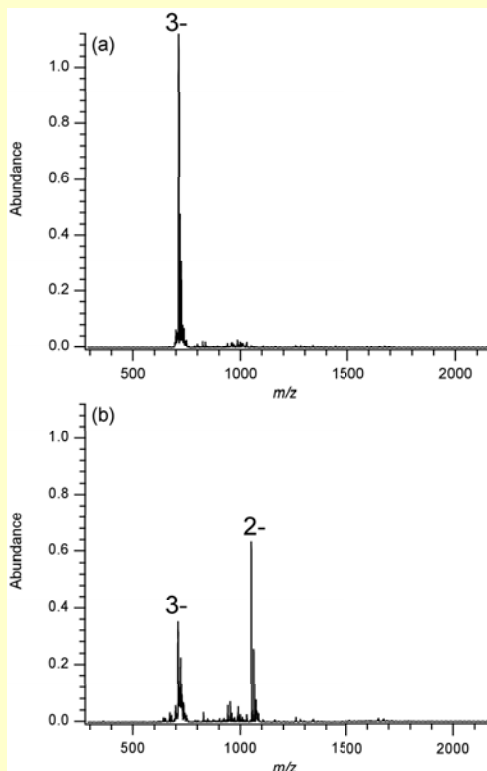


Rates of  $4- \rightarrow 3-$  and  $3- \rightarrow 2-$  for  $T_7$ -dye as  $f(T)$

# RESULTS

## Charge state conversion of underivatized oligonucleotides

- Process of  $M^{n-} \rightarrow M^{(n-1)-}$  also observed from underivatized oligonucleotides (see  $A_7^{3-} \rightarrow A_7^{2-}$  below)

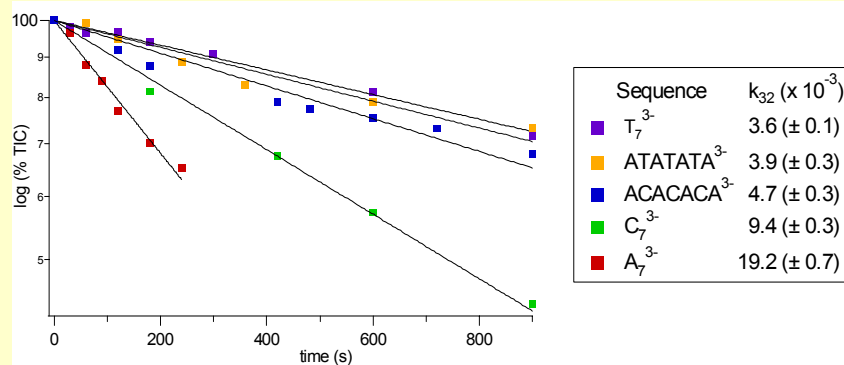


Mass spectra of  $(A_7)^{3-}$  at varied heating times at 102°C.  
(a) 0 s (isolation of 3- charge state); (b) 240 s.

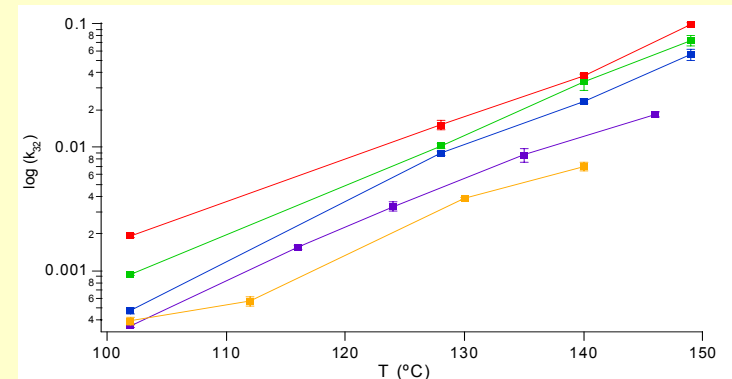
$3- \rightarrow 2-$  for  $A_7$  as  $f(t)$  at 102°C

# RESULTS

- All 7-mer oligonucleotides studied underwent charge state conversion
- For most sequences (except ATATATA), charge state conversion was only process observed
- Rates of charge state conversion varied with sequence
- Sequence dependence persisted over T range



3- → 2- for various oligonucleotide 7-mers as f(t) at 102°C



3- → 2- rates for 7-mers as f(T) (see legend above left)

# RESULTS

## Electron autodetachment

- $M^{n-} \rightarrow M^{(n-1)-}$  phenomenon proposed to occur via electron autodetachment
- Electron autodetachment occurs due to Coulomb repulsion in multiply charged anions
  - In small rigid molecules, has been shown to vary for different structural isomers having different electron charge separations [2, 3]
  - Has been shown to occur following vibrational heating via multiphoton infrared absorption [4]
  - Theoretical calculations describe transfer of vibration to electronic energy [5]
- Electron autodetachment from oligonucleotides dominated by conformation-induced changes in charge separation
  - Single-stranded oligonucleotide conformations very flexible and vary as a function of
    - charge state
    - temperature
    - base composition

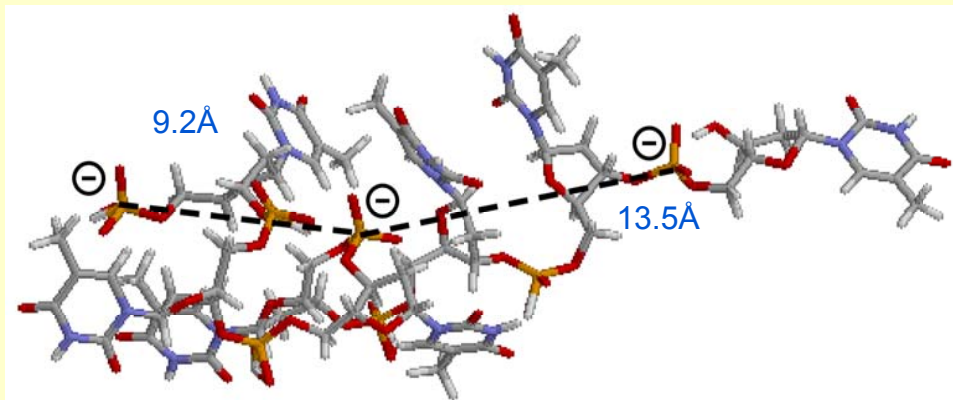
# RESULTS

- Other dissociation or ion/molecule reactions are not likely causes of observation of  $\mathbf{M}^{n-} \rightarrow \mathbf{M}^{(n-1)-}$  because
  - Total ion population conserved
  - Likely product ions from dissociation of oligonucleotides would appear above cutoff  $m/z$  of  $\sim 300$  Da
  - No trend observed in variation of rates of charge loss as a function of He pressure
  - RGA measurements indicate no contaminants present and no change in background gases except  $\sim 3x$  increase in water vapor from room to elevated temperatures

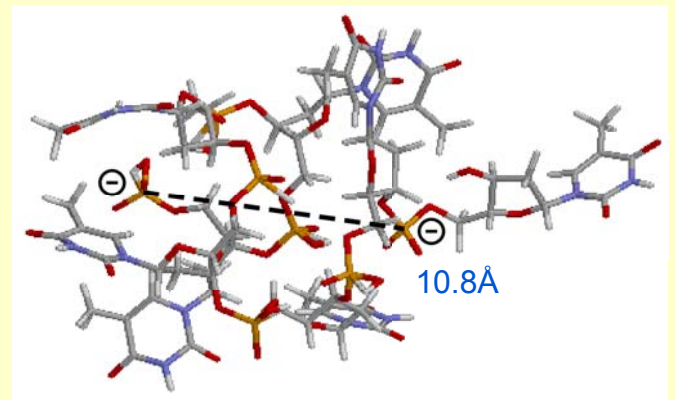
# RESULTS

## Molecular dynamics simulations

- Initial calculations of  $(T_7)^{3-}$  and  $(T_7)^{2-}$  performed [6]
  - Insight II software, 298 K, extensible systematic forcefield optimized for vacuum
  - Negative charges localized on phosphate groups
  - Structures illustrate flexibility of single-stranded oligonucleotides



$T_7^{3-}$



$T_7^{2-}$

# SUMMARY

## Understanding parameters affecting electron autodetachment rates

- Fluctuations in conformation
  - increased fluctuations → decreased <charge separation> → decreased detachment energy barrier
- Charge State
  - increased charge state → increased Coulomb repulsion
- Temperature
  - increased population of higher vibrational modes → increased flexibility and increased probability of electronic curve-crossing
  - increased internal energy (RRKM) may contribute energy to excitation of vibrational modes
- Sequence
  - Flexibility of oligonucleotides correlated with base stacking interactions [7]
  - Dielectric screening of Coulomb fields vary with polarizability of nucleotides
- Molecular dynamics simulations required for further understanding

## REFERENCES

- [1] Danell, A. S.; Parks, J. H. *Int. J. Mass Spectrom.* **2003**, *in press*.  
For more information on FRET experiments, see TPK # 197.
- [2] Wang, L.-S.; Wang, X.-B. *J. Phys. Chem. A*, **2000**, *104*, 1978-1990.
- [3] Wang, X.-B.; Nicholas, J. B.; Wang, L.-S. *J. Chem. Phys.* **2000**, *113*, 653-661.
- [4] Wight, C. A.; Beauchamp, J. L. *J. Am. Chem. Soc.* **1981**, *103*, 6499- 6501.
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- [6] Wenzel, W.; Merlitz, H. Personal Communication, **2003**.
- [7] Goddard, N. L.; Bonnet, G.; Krichevsky, O.; Libchaber, A. *Phys. Rev. Lett.* **2000**, *85*, 2400-2403.